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Introduction; 18.1 Periodicity; 18.2 Occurrence and Preparation of the Representative Metals; 18.3 Structure and General Properties of the Metalloids; 18.4 Structure and General Properties of the Nonmetals; 18.5 Occurrence, Preparation, and Compounds of Hydrogen; 18.6 Occurrence, Preparation, and Properties of Carbonates; 18.7 Occurrence, Preparation, and Properties of Nitrogen

Answer Key Chapter 4 - Chemistry 2e | OpenStax

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4.3. For a reaction, $A + B \rightarrow \text{Product}$; the rate law is given by, $r = k [A]^{1/2} [B]^2$. What is the order of the reaction? Ans. Order of reaction. = $1/2 + 2 = 2\ 1/2$ or 2.5. 4.4. The conversion of molecules X to Y follows second order kinetics. If concentration of X is increased to three times how will it affect the rate of formation of Y ?

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Chemistry Chapter 4 Atomic Structure Answer Key

Answer: According to VSEPR theory, if CH₄ were square planar, the bond angle would be 90°. For tetrahedral structure, the bond angle is 109°28'. For tetrahedral structure, the bond angle is 109°28'.

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Question 1

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