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Kinetics And Catalysis In Microheterogeneous

This book provides a quantitative assessment of the advances in the area of catalysis and kinetics in microheterogeneous systems. It is an invaluable resource for chemists interested in catalysis and reaction kinetics, and physicists interested in semiconductors, metal clusters and catalysis.

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This book consists of two parts: the first one deals with kinetics of processes in heterogeneous systems and those with different degrees of phase dispersion. Strict analytic methods are recommended for solving the kinetic equations for non-catalytic processes in gas, liquid, and solid phases herein.

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Kinetics and Catalysis Russian is a periodical that publishes theoretical and experimental works on homogeneous and heterogeneous kinetics and catalysis. Other topics include the mechanism and kinetics of noncatalytic processes in gaseous, liquid, and solid phases, quantum chemical calculations in kinetics and catalysis, methods of studying catalytic processes and catalysts, the chemistry of catalysts and adsorbent surfaces, the structure and physicochemical properties of catalysts ...

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Heterogeneous Catalysis. In heterogeneous catalysis, the catalyst is in a different phase from the reactants. At least one of the reactants interacts with the solid surface in a physical process called adsorption in such a way that a chemical bond in the reactant becomes weak and then breaks.

18.7: Kinetics of Catalysis - Chemistry LibreTexts

The saturation parameter compares the value of the substrate concentration in the layer to the Michaelis constant. When this parameter is small the catalytic kinetics is unsaturated and the rate is first order with respect to substrate concentration. When it is large the kinetics are saturated and zero order kinetics pertain.

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